

Extension of Compaction Bands in Porous Sandstones

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Collaborators

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Support: U.S. Dept of Energy,
Office of Basic Energy Sciences

What do compaction bands look
like in the field?

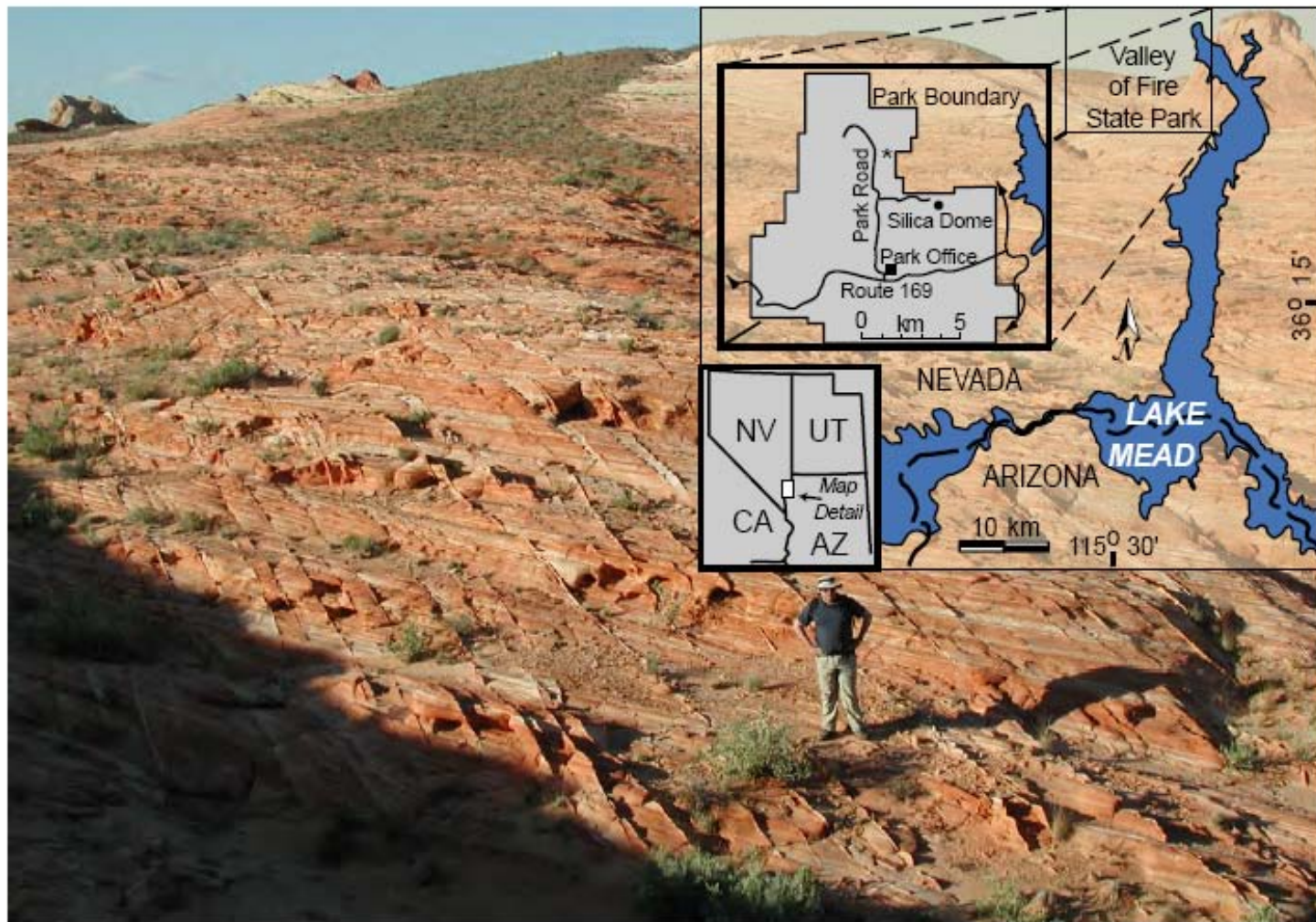


Figure 1. Location of the Valley of Fire State Park, southeastern Nevada (inset), where more than 20 square kilometers of the 1,400-m-thick eolian Jurassic Aztec Sandstone are extensively exposed. Photo shows view northward from the location marked * on the park detail. Throughout the upper half of the Aztec, compaction bands crop out in positive relief as sub-parallel, centimeter-thick, north-northwest-trending, steeply east-dipping tabular fins spaced from centimeters to meters apart.

Compaction Bands: Earliest Structural Fabric of the Aztec

From Kurt Sternlof, Stanford

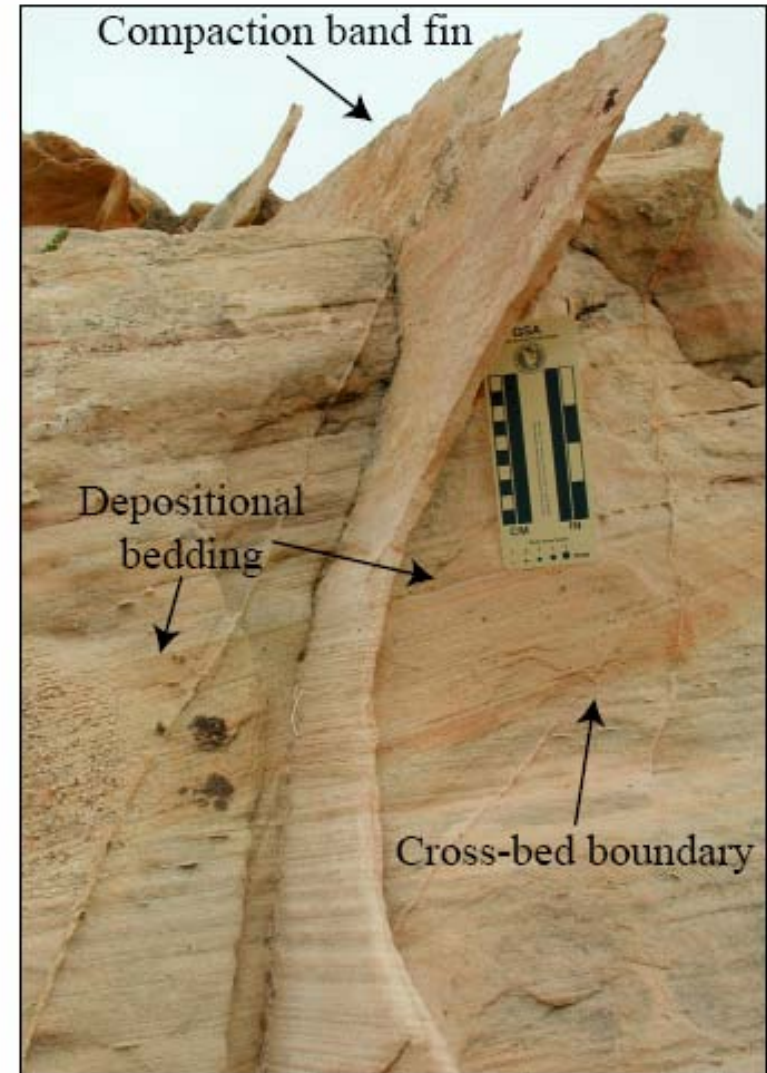


Figure 3. Close-up of a typical, well-developed compaction band fin in outcrop. Note that depositional bedding extends relatively undisturbed across the band, and is clearly visible on the fin.

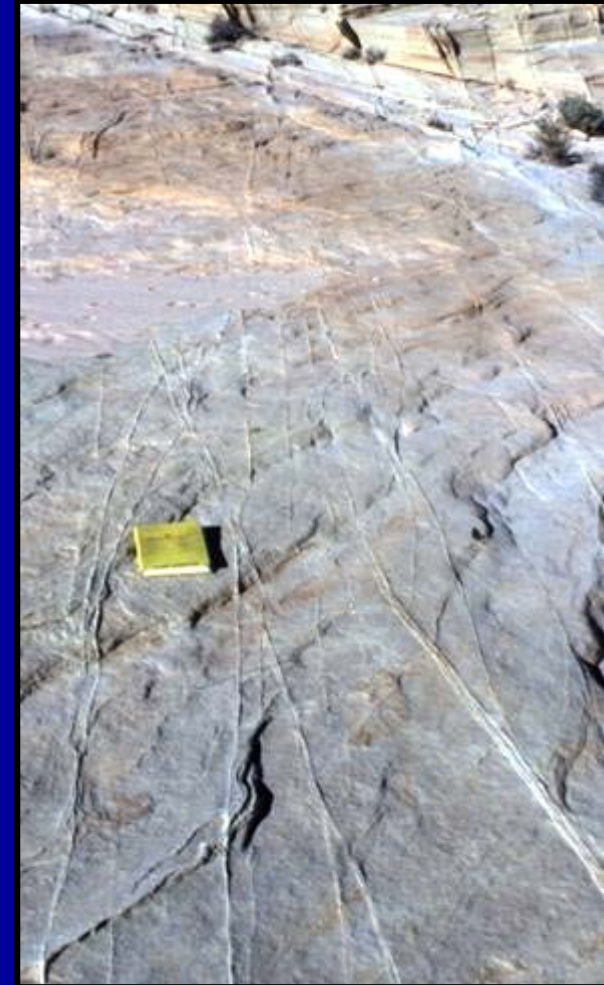
Characteristic Compaction Band Arrays



Parallel



Cross-hatch



Anastomosing

in the Aztec Sandstone

From Kurt Sternlof, Stanford

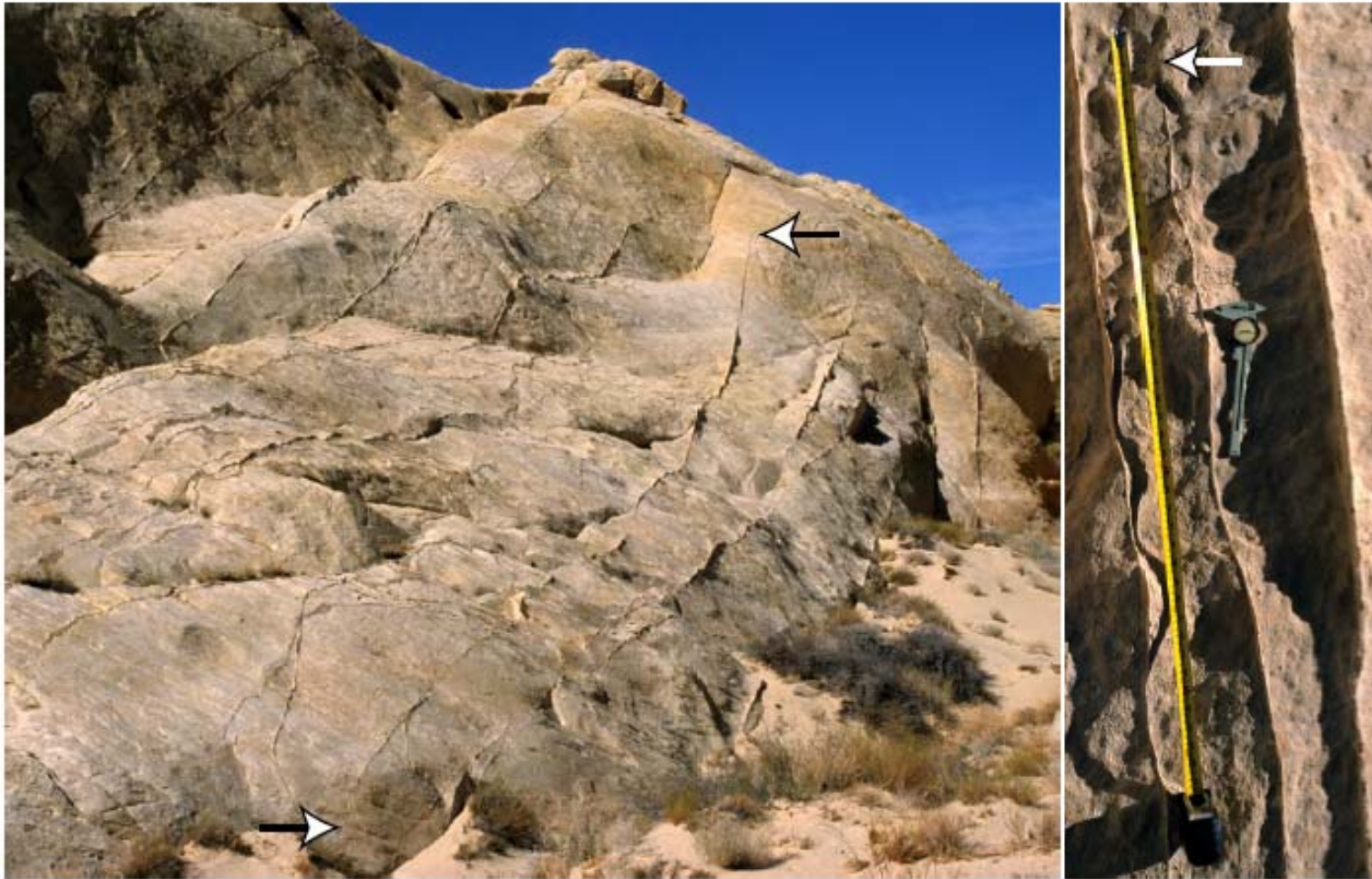


Figure 2. Left-hand photo shows an outcrop of widely spaced, relatively planar and parallel compaction bands along the northeast flank of Silica Dome. Arrows indicate opposite tips of a single band 62 m long and up to 15 mm thick (illusory gaps in band continuity are due to outcrop topography and breaks in the telltale fin). A total of 16 tip-to-tip thickness profiles were measured using a steel tape and calipers. Right-hand photo illustrates that, even when closely spaced, compaction bands in this locale tend to remain planar (arrow indicates tip).

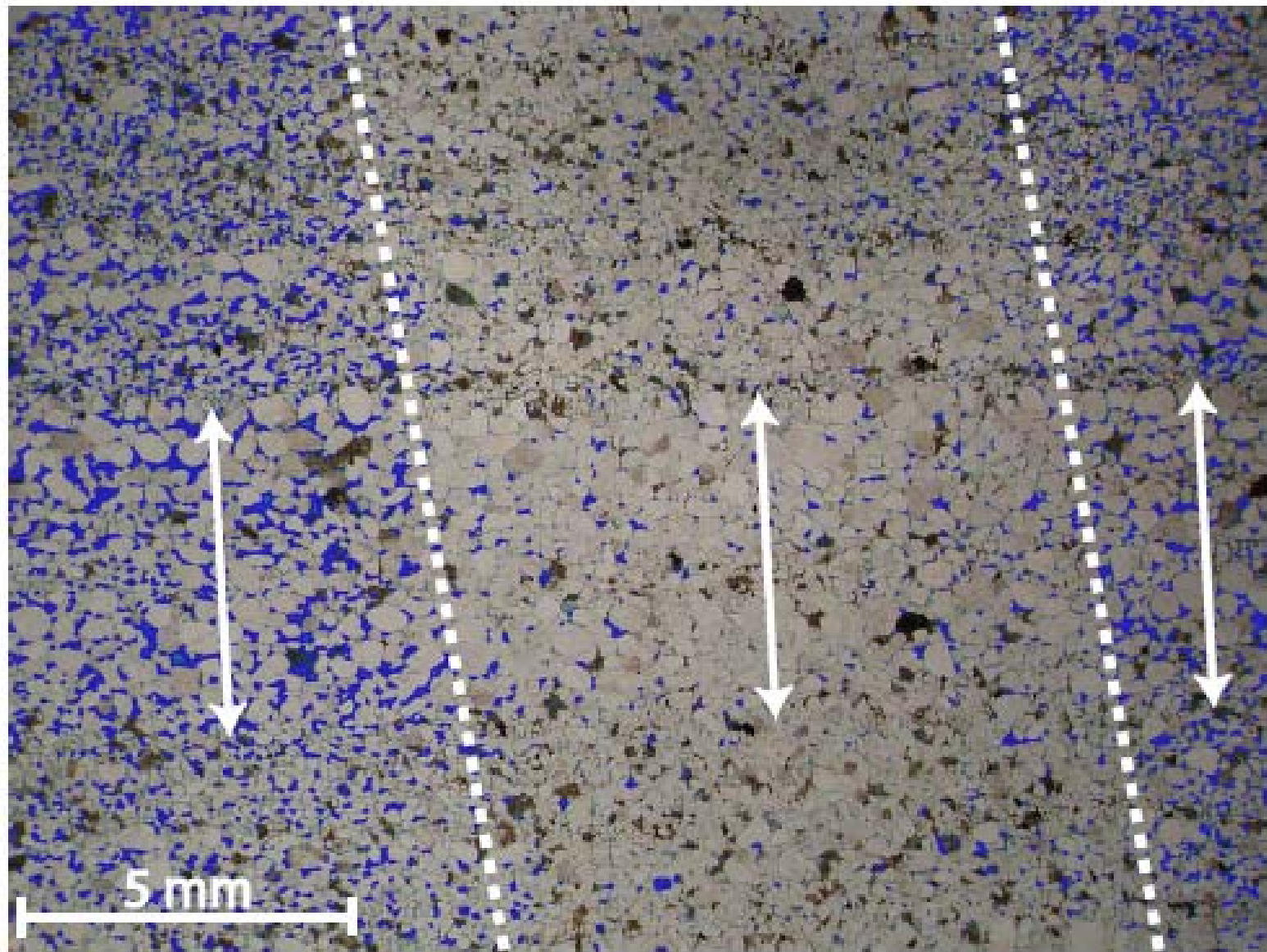


Figure 5. Photomicrograph of compaction band sampled 6.0 meters from the tip, where it is about 9 mm thick (dotted black lines). Blue indicates epoxy-filled pore space (~ 25% outside the band, ~ 10%

From Sternlof, Rudnicki and Pollard, JGR, 2005

Earth a solution to air pollution?

Scientists seriously consider injecting gases in ground

By Julie Deardorff
Tribune staff reporter

The plan to landfill air pollution might seem laughable.

As a stopgap solution to global warming, scientists have proposed capturing several billion tons of carbon dioxide from the air and injecting it deep into the earth for long-term storage.

No one knows whether vast amounts of the greenhouse gas would stay put 2 miles below ground. Nevertheless, an increasing number of experts—including some environmentalists—believe the idea isn't as harebrained as it might sound.

With carbon dioxide emissions rising steadily in the U.S. and around the world, countries are casting about for ways to reduce the heat-trapping pollution. In the meantime, scientists say it can be unloaded into dark reaches of the earth, including saline aquifers, depleted oil wells, coal seams and the ocean.

The sprawling Illinois Basin, which extends into Indiana and western Kentucky, offers an ideal location to study three of the methods, say Illinois State Geological Survey officials. They are leading a multistate effort to bring up to \$10 million in federal funding to the region to study and, perhaps, begin testing the technique.

Last month, the U.S. Department of Energy expanded funding to inspire state agencies, in-

PLEASE SEE **GASES**, BACK PAGE

Carbon dioxide 'injection' could reduce greenhouse effect

In an attempt to combat global warming, scientists are exploring whether it's possible to inject carbon dioxide emissions directly from power plants deep into the ground. Carbon dioxide would be captured from a power plant, transformed into liquid and pipelined through injection wells drilled into geologic formations.

INJECTED TO PRODUCE MORE RESOURCES

Coal bed (1,000-1,500 ft.)
A chemical bond between coal and carbon dioxide would displace the methane in coal. Methane then would be pumped out to power plants, fuel cells and gas turbines to produce more electricity. Some carbon dioxide would be stored in the coal.

Oil reservoir (1,500-2,500 ft.)
Carbon dioxide would thin oil and replenish energy in the reservoir, making it easier for oil to be pushed out and recovered. Leftover carbon dioxide would remain inert and be stored in the reservoir.

ENERGY TO CONSUMER

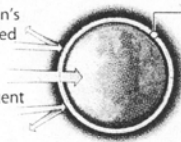
INJECTED FOR STORAGE

Saline formation (5,000 ft. or deeper)
Saline would offer a larger storage capacity than coal or oil reservoirs. It can store up to several hundred years worth of carbon dioxide emissions.

GREENHOUSE EFFECT AND GLOBAL WARMING

Most of the sun's energy is absorbed by the Earth's surface.

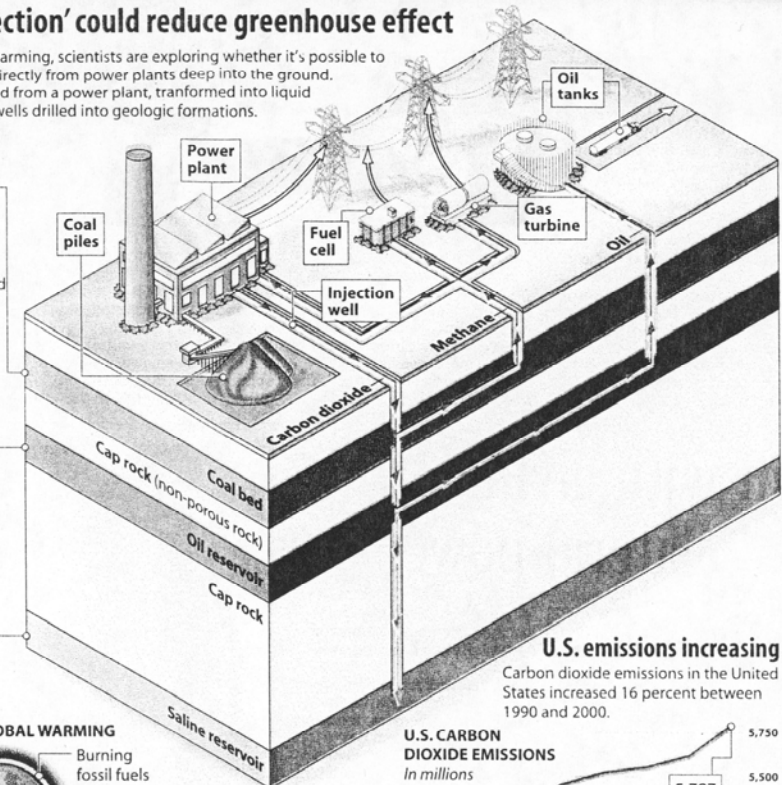
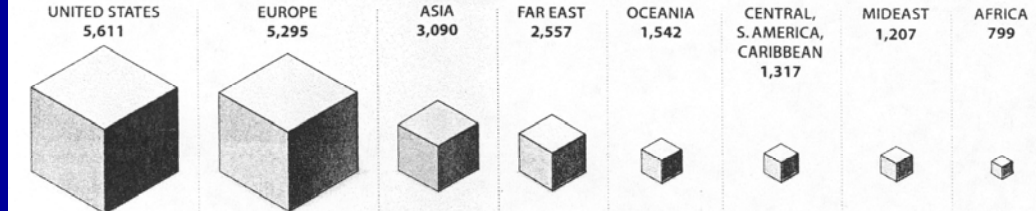
About 30 percent of the energy is reflected.



Burning fossil fuels increases carbon dioxide in the atmosphere. The gas forms a blanket, which traps heat and warms the Earth.

INTERNATIONAL CARBON DIOXIDE EMISSIONS

In millions of metric tons, 1999

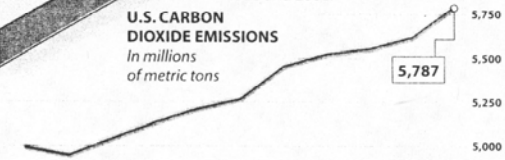


U.S. emissions increasing

Carbon dioxide emissions in the United States increased 16 percent between 1990 and 2000.

U.S. CARBON DIOXIDE EMISSIONS

In millions of metric tons



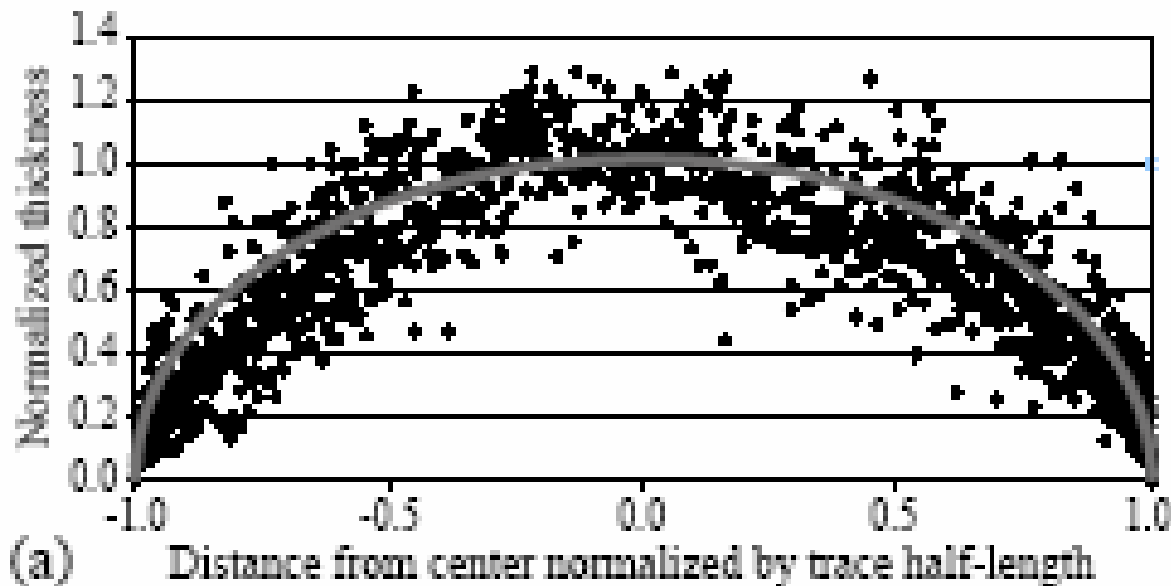
Chicago Tribune
Front Page,
January 27, 2003

In June, Ronald Oxburgh, Shell's chief in the United Kingdom called sequestration essentially the last best hope to combat climate change. “If we don't have sequestration, then I see very little hope for the world, Oxburgh told the British newspaper *The Guardian*.”

Science, August 13, 2004.

Normalized band thickness vs. distance from center normalized by half-length.

From Sternlof, Rudnicki and Pollard (JGR, 2005)
and Sternlof (PhD Thesis, 2006)

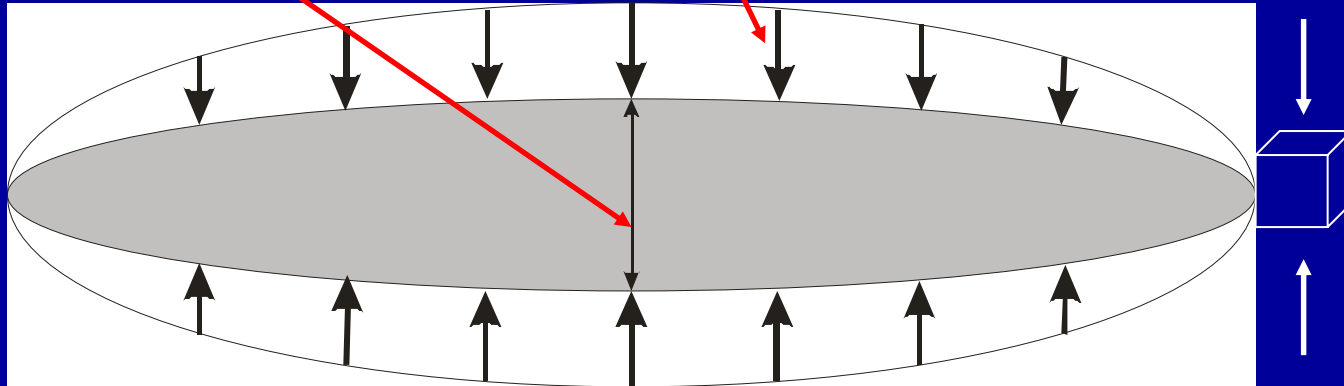


Ellipsoidal band with uniform inelastic compactive strain ε^p

$$\text{compactive displacement} = \frac{w\varepsilon^p}{1 - \varepsilon^p} \approx w\varepsilon^p$$

$2w =$ band width

compressive stress ahead of the band is elevated sufficiently to cause compaction



Special (Relevant) Case

Uniaxial farfield strain: $\varepsilon_{11}^{\infty} = \varepsilon_{22}^{\infty} = 0$

$$\varepsilon_{33}^{\text{tip}} = \varepsilon_{33}^{\text{band}}$$

For very small aspect ratios

(: 10^{-3} to 10^{-4} from Sternlof data):

$$\sigma_{33}^{\text{band}} ; \sigma_{33}^{\infty}$$

$$\varepsilon_{11}^{\infty} = \varepsilon_{22}^{\infty} ; 0$$

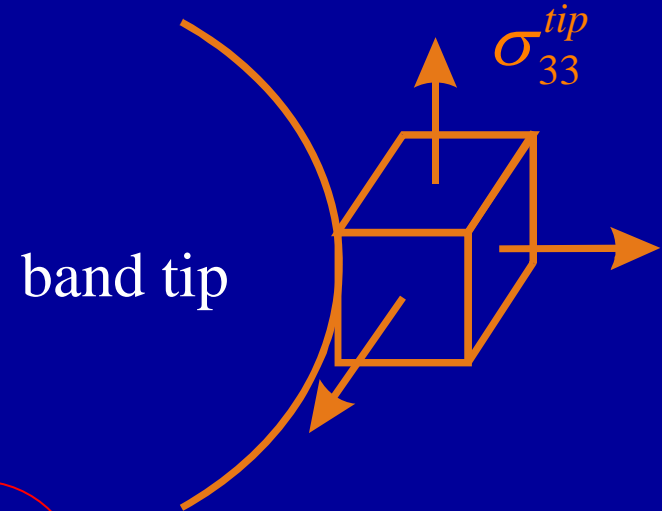
and at tip:

$$\varepsilon_{33}^{\text{tip}} = \varepsilon_{33}^{\text{band}} \Rightarrow \sigma_{33}^{\text{tip}} = \sigma_{33}^{\text{band}} \frac{M_{\text{band}}}{M} + M \varepsilon_{33}^p$$

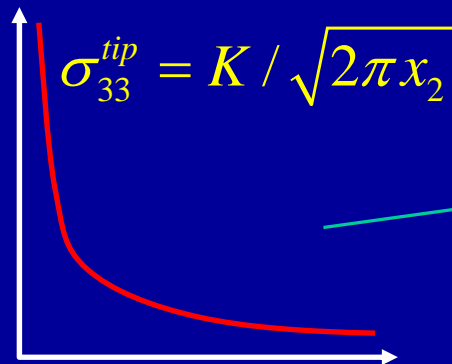
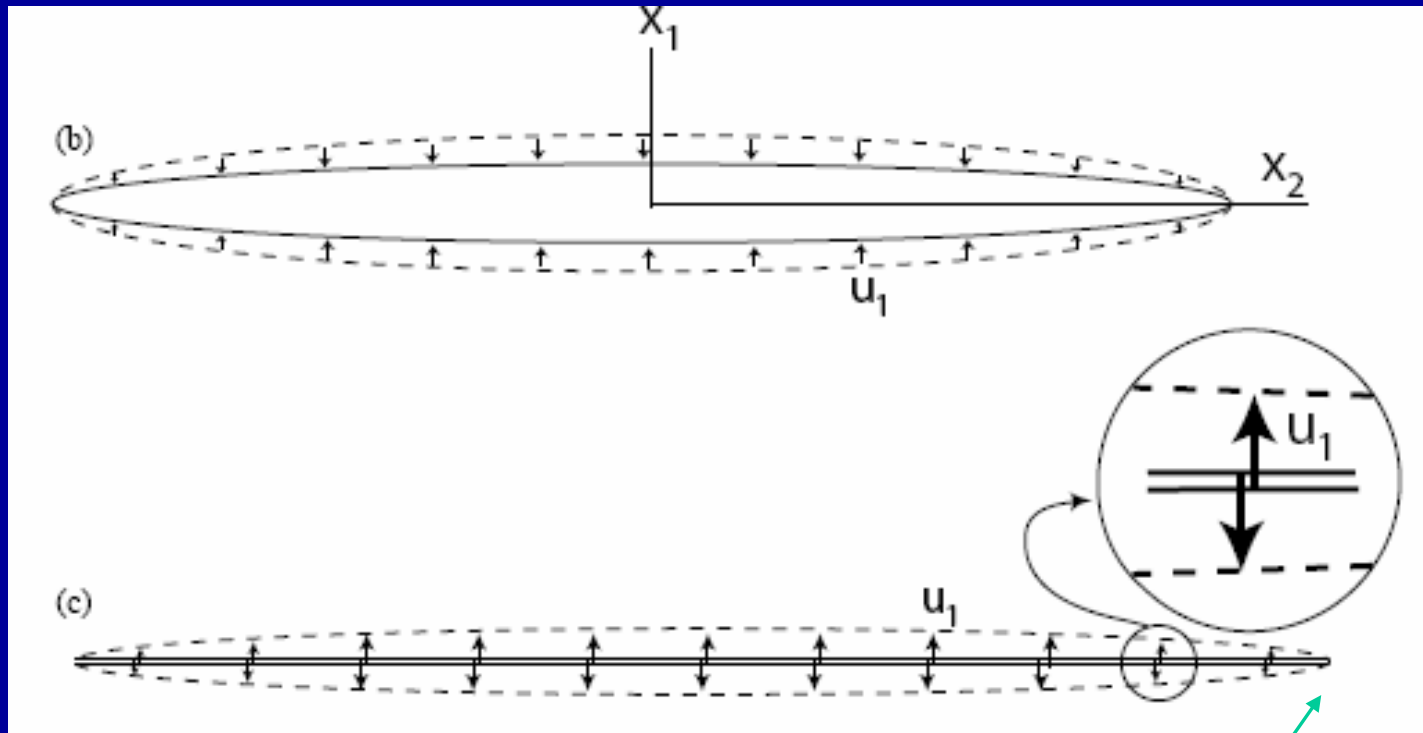
$$\sigma_{33}^{\infty} : 40 \text{ MPa}$$

$$M = 2\mu(1-\nu)/(1-2\nu) \approx 40 \text{ to } 50 \times \sigma_{33}^{\infty}$$

(isotropic elasticity)



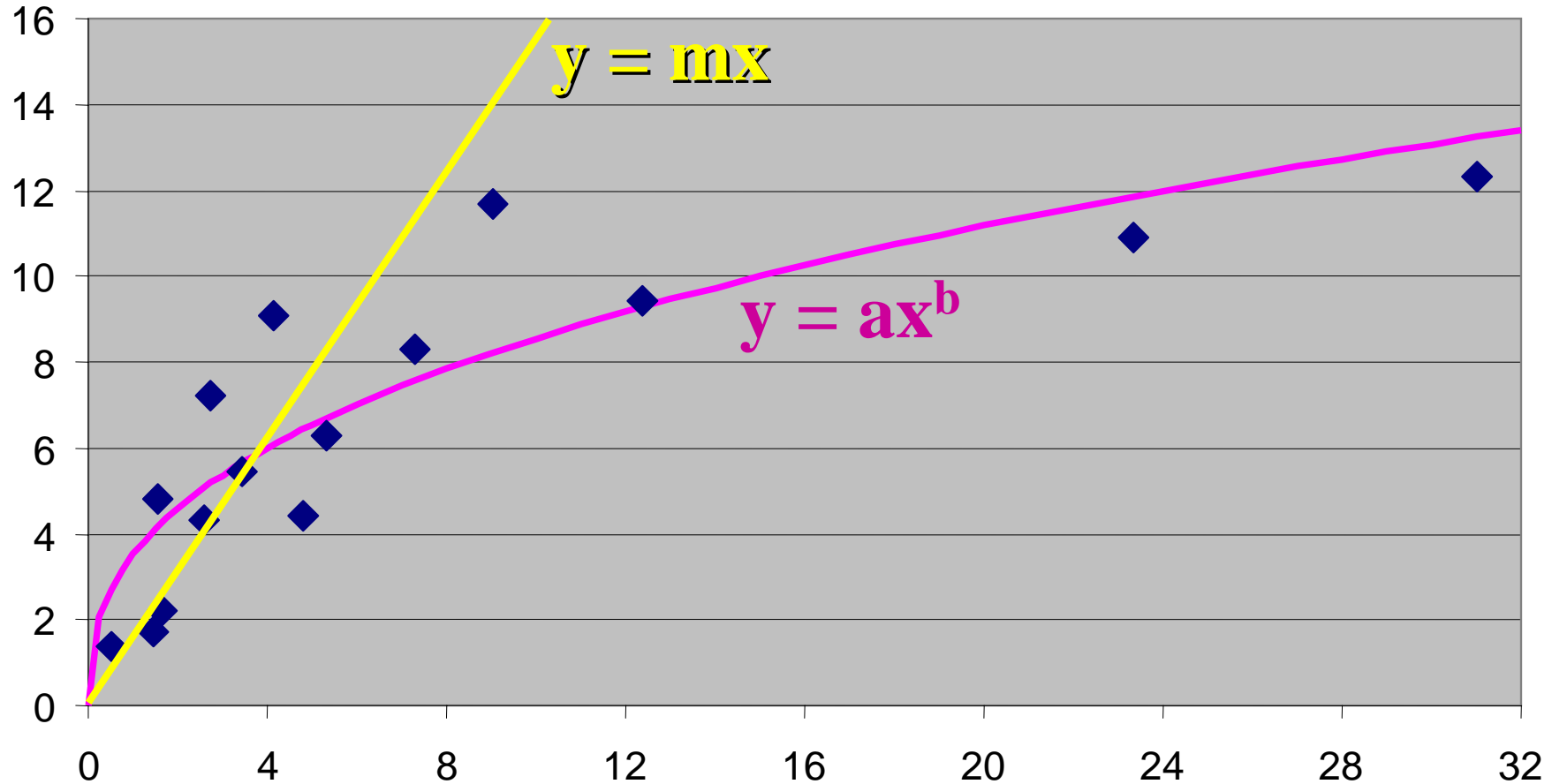
“Anti-crack model” (Pollard, Fletcher, Sternlof)



Compaction Bands as Stiff Inclusions

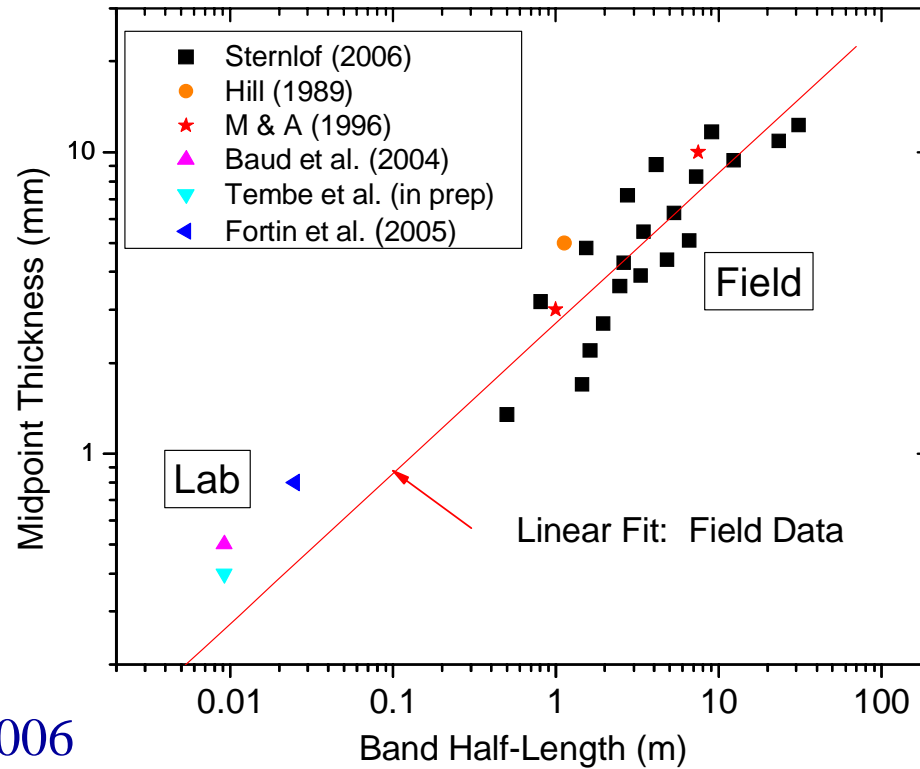
From Sternlof

Band width (mm)



Band half length (m)

Figure 1: Compaction Band Data

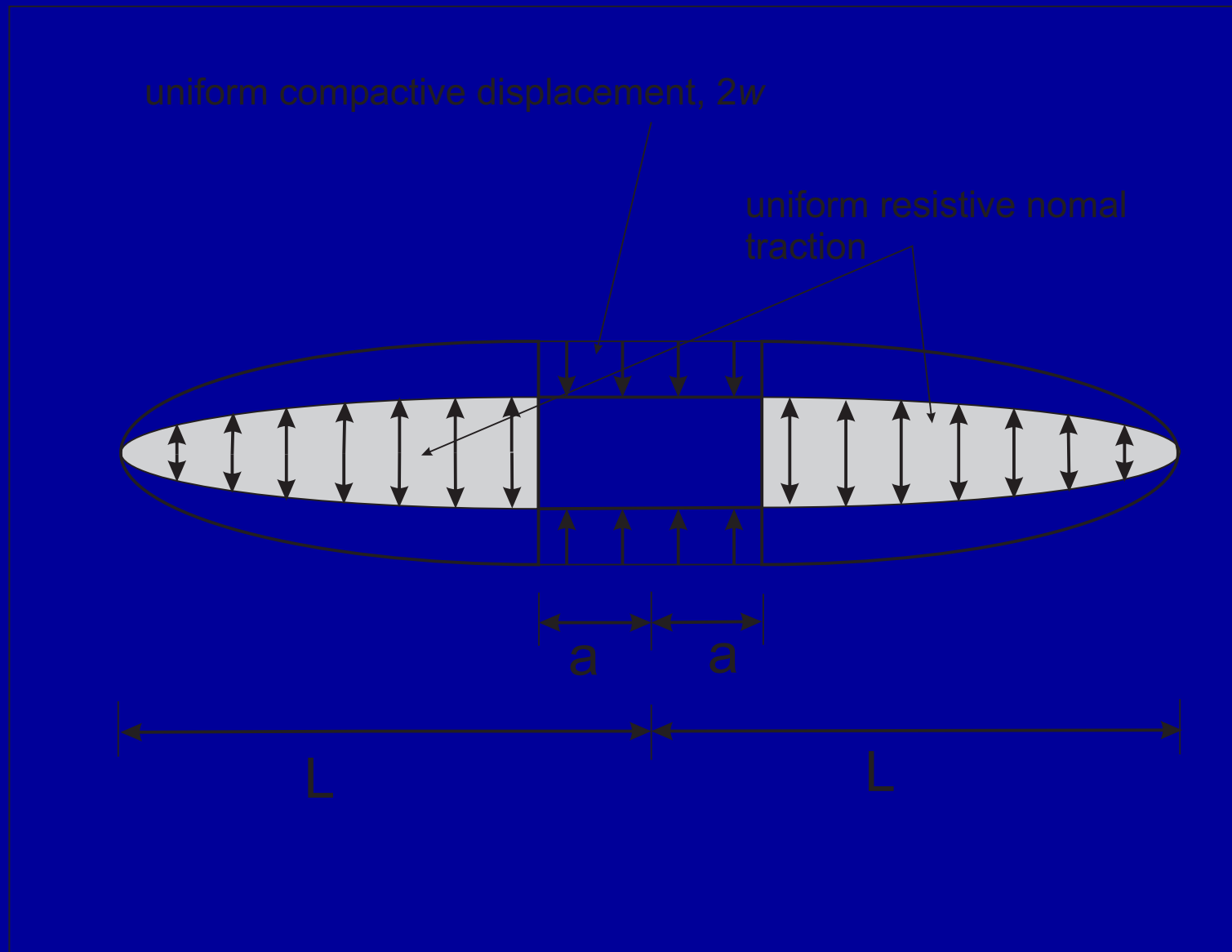


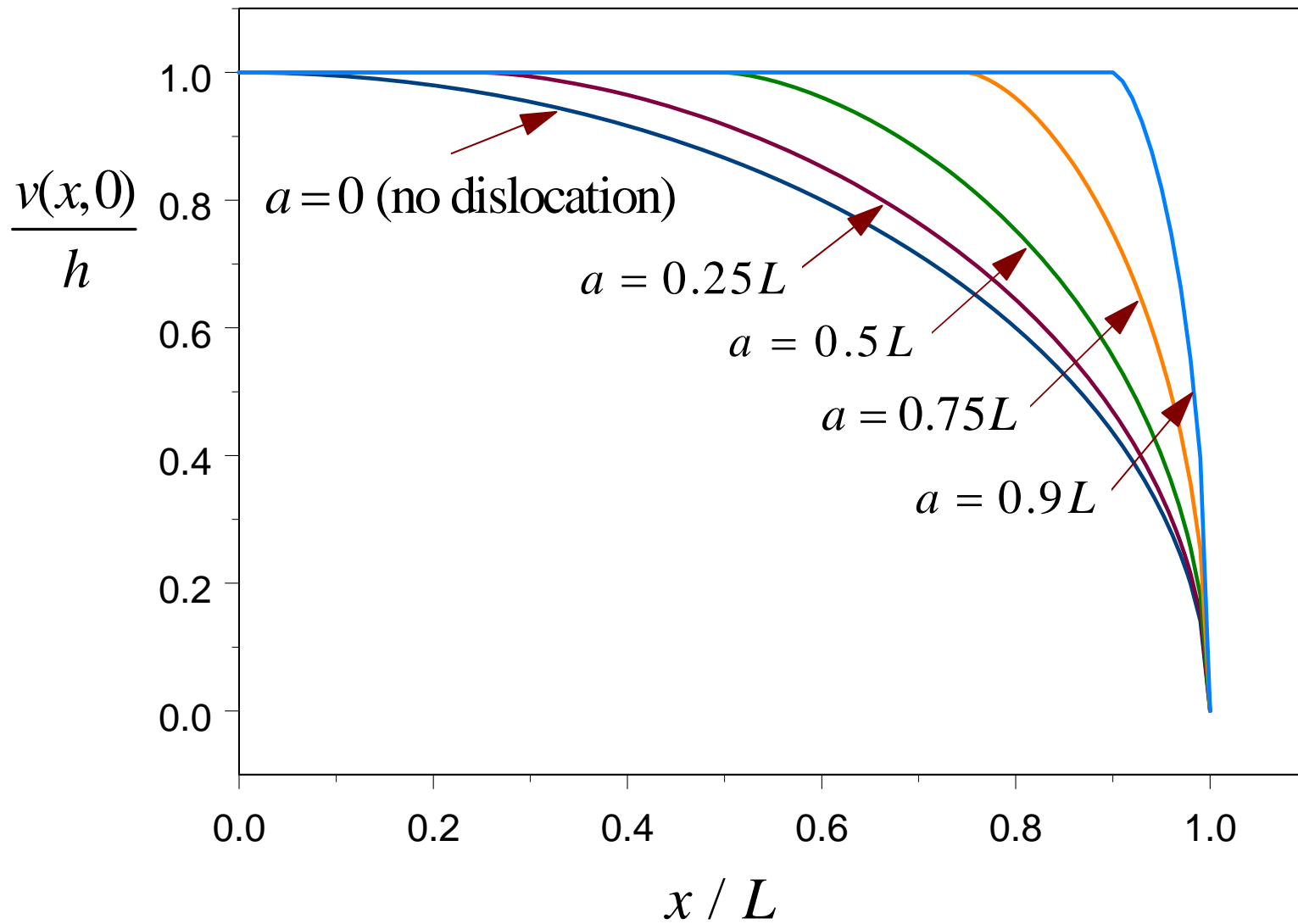
After Tembe, 2006

Linear Fit: $y = A + Bx$

	A	B
Field Data	0.43240	0.49891

Combined “anti-crack” and “anti-dislocation” model





$$K = \frac{\mu w}{(1-\nu)} \sqrt{\frac{\pi}{L-a}} \left[\frac{k \sqrt{L-a}}{\frac{E(k)}{4} - \frac{(1-k^2)K(k)}{2}} \right]$$

$F(k)$

where $k^2 = 1 - (a/L)^2$

$$G_{\text{crit}} = \frac{(1-\nu)}{2\mu} K_{\text{crit}}^2 \Rightarrow K_{\text{crit}} = \sqrt{\frac{2\mu G_{\text{crit}}}{(1-\nu)}}$$

$F(k) = 0.924 \text{ for } L = 3a$
 $F(k) = 0.942 \text{ for } L = 5a$

$$\therefore 2w = \sqrt{L} \sqrt{\frac{8}{\pi} \frac{(1-\nu)G_{\text{crit}}}{\mu}} \underbrace{F(k)}_{\approx 1}, \quad L \gg a$$

$$2w ; \sqrt{L} \sqrt{\frac{8}{\pi} \frac{(1-\nu)G_{\text{crit}}}{\mu}}, L \gg a$$

For $\nu=0.2$, $\mu = 8.333$ GPa,

Fit to $2w$ vs L data $\Rightarrow G_{\text{crit}} = 30 \text{ kJ/m}^2$

From Sternlof and Rudnicki, GRL, 2005:

critical energy release rate

$$G_{\text{crit}} ; \underbrace{h\xi\varepsilon^p}_{\text{net compaction}} \underbrace{\frac{M(\Delta/h)}{14 \cdot 2 \cdot 4}}_{\text{stress}=\sigma^+} + O(\xi^2)$$

$$\sigma^+ = 40 \text{ MPa}$$

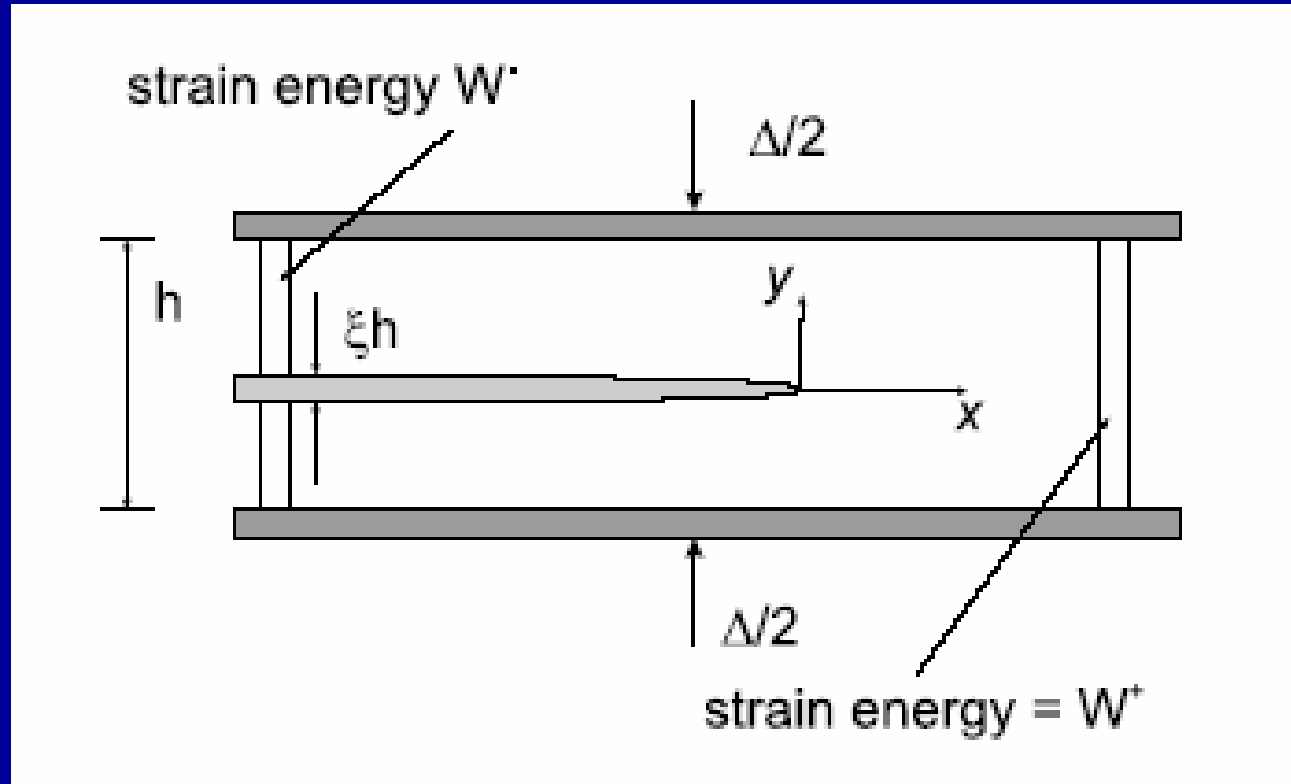
$$h\xi = 0.01 \text{ (1 cm thick CB, spaced 1 meter apart)}$$

$$\varepsilon^p = 0.1 \text{ (corresponding to 10\% porosity reduction)}$$

$$\Rightarrow G = 40 \text{ kJ/m}^2 \text{ (10 to 60 kJ/m}^2 \text{ for range of stress estimates)}$$

Energy Release Due to Compaction Band Propagation

Rudnicki and Sternlof, GRL, 2005 (after J. R. Rice, 1968)



energy release rate $G = W_+ - W_-$

$$= \frac{1}{2} \frac{Mh}{(M/M_b)\xi + (1-\xi)} \left\{ \left(\frac{\Delta}{h} \right)^2 \xi \left(\frac{M}{M_b} - 1 \right) + 2\xi \varepsilon^p \left(\frac{\Delta}{h} \right) - \left(\xi \varepsilon^p \right)^2 \right\}$$

Tembe et al., JGR, 2006 estimate “compaction energies” (equivalent to energy release rate) of **16 – 43 kJ/m²** for Berea and Bentheim sandstones.

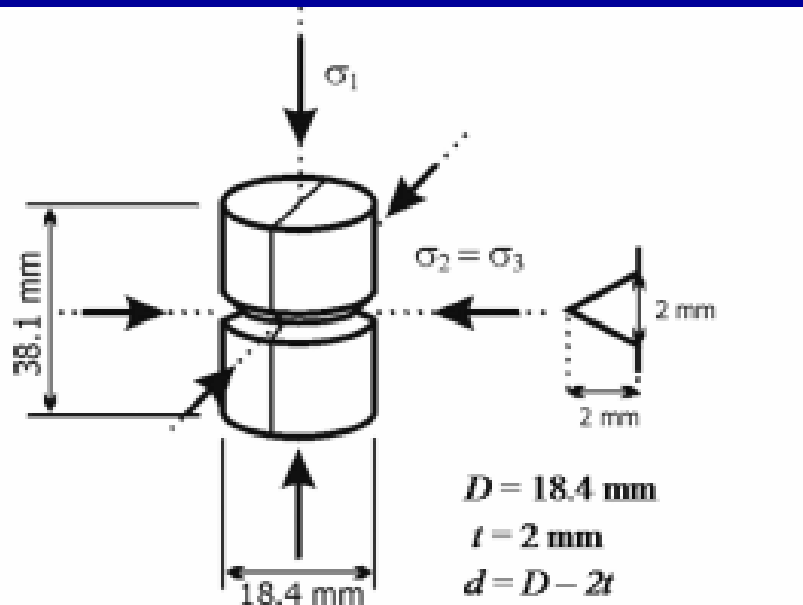
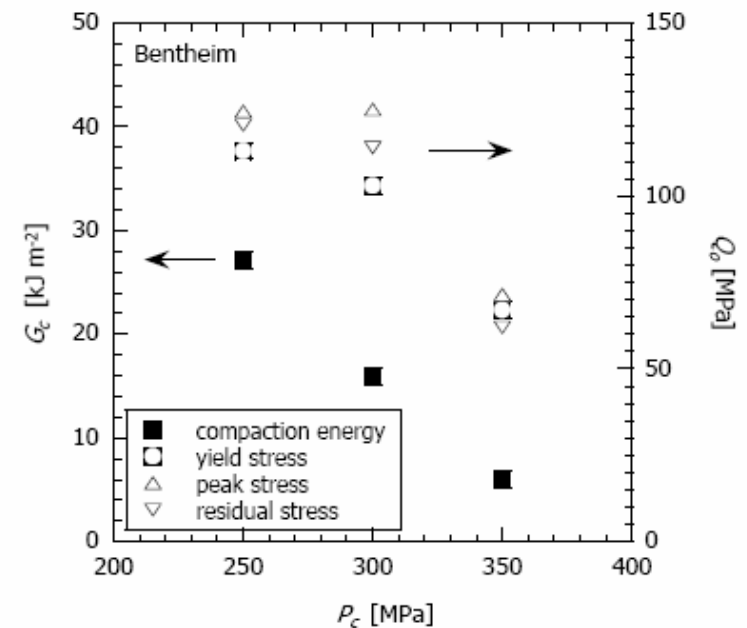


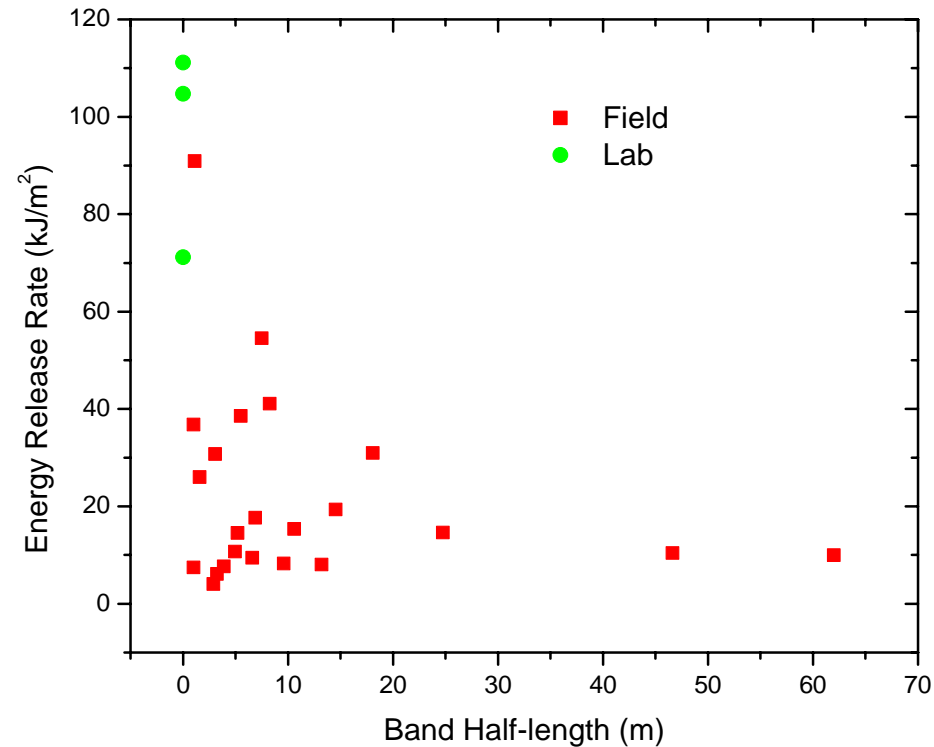
Figure 1. Geometric configuration of a notched sample loaded in a conventional triaxial apparatus.



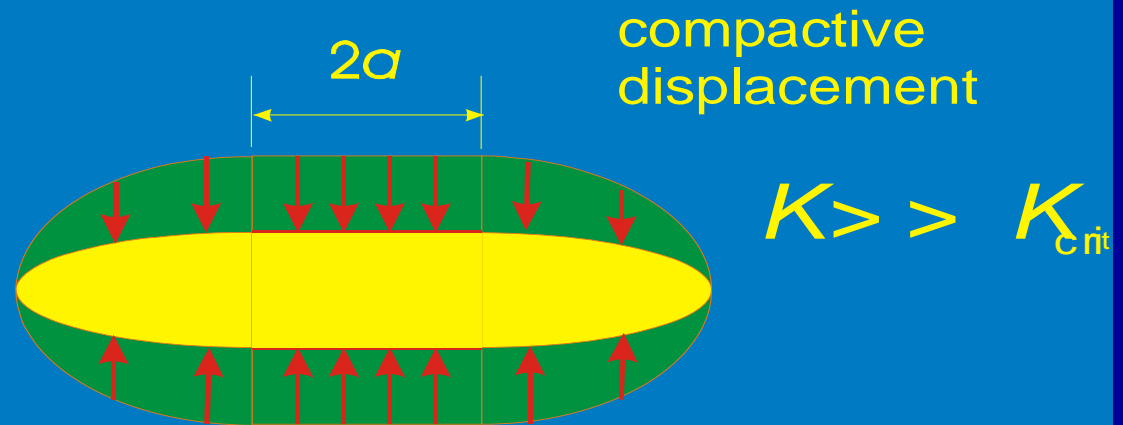
Energy release rate G_{crit} predicted from

$$2w = \sqrt{L} \sqrt{\frac{8}{\pi} \frac{(1-\nu)G_{\text{crit}}}{\mu}} F(k) \quad \left(\frac{L}{2w} \approx 1 \right)$$

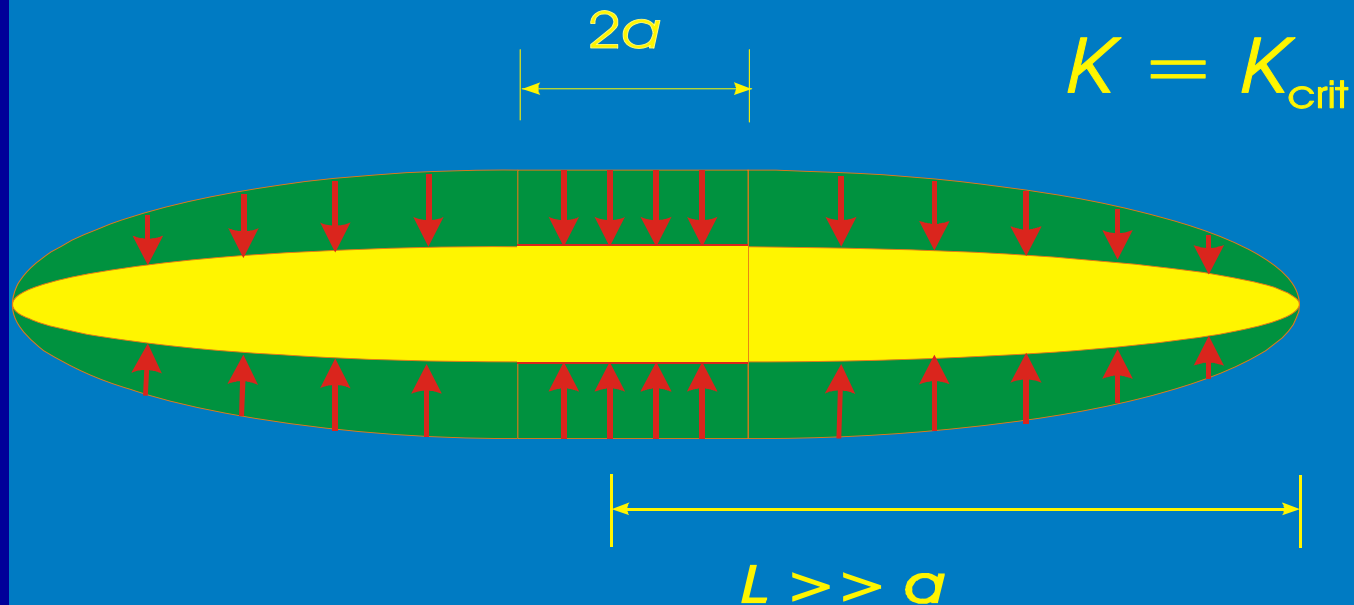
with Sternlof's data for $2w$ and L ,
 $\nu=0.2$, $\mu = 8.333$ GPa.



Initial compactive event

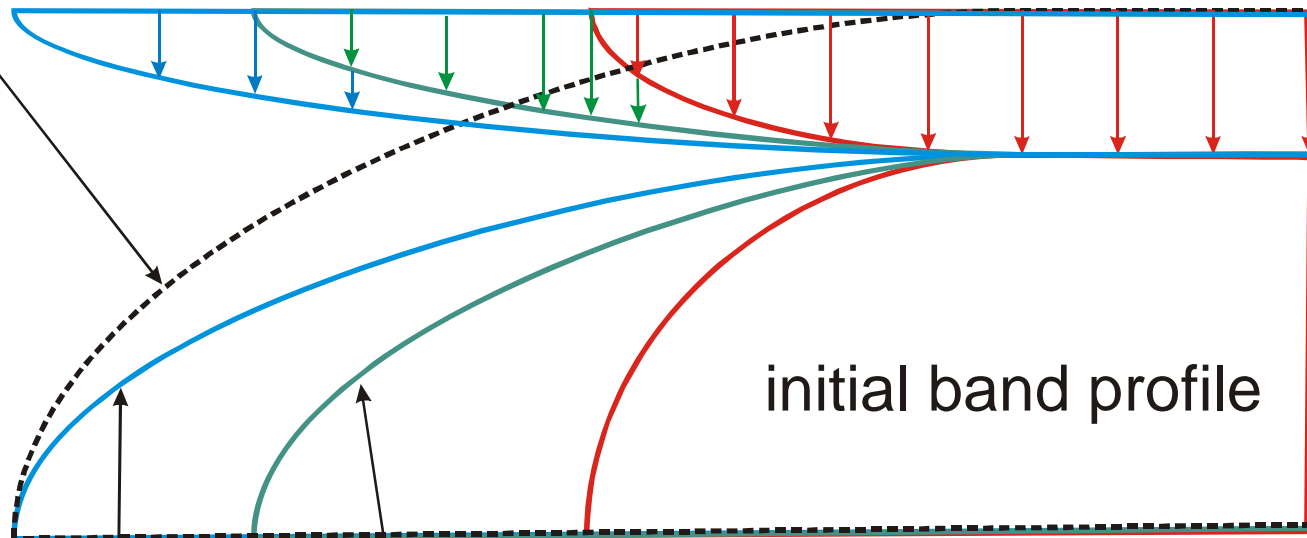


Final band length



initial position of material
forming final band boundary

compactive displacements



initial band profile

intermediate band profile

final band profile

Conclusions

- A simple fracture and dislocation model indicates a (surprising) consistency between lab and field data (though multiple interpretations are possible).
- More field and laboratory observations of the breakdown process near the tip of bands would be helpful in constructing more elaborate models.

Thank You!